



E. I. DU PONT DE NEMOURS & COMPANY
INCORPORATED

WILMINGTON, DELAWARE 19898

ENGINEERING DEPARTMENT

42966
CC: C.M. Hunter-Photo Prod.- Wilmington
M.L. Ward-Photo Prod.-Wilmington
J.E. Dickens-Photo Prod.-Exp. Sta.
W.G. Bottjer-Photo Prod.-Exp. Sta.
R.F. Bree-Photo Prod. - Newport
E.H. Kolb -Pigments - Newport
R.M. Haire - Pigments- Newport
G.A. Christian
A.C. Barlow
R.G. Kissell, Jr.
J.L. Parsons
RC 13/AENSW

February 10, 1970

W. B. JENNINGS
PHOTO PRODUCTS DEPARTMENT
NEWPORT PLANT

PHOTO PRODUCTS DEPARTMENT - NEWPORT PLANT
SOLID WASTE DISPOSAL
MAGNETIC TAPE VENTURE - WASTE TAPE DISPOSAL

At your request, I have studied the problems associated with disposal of waste Crolyn[®] tape including the adequacy of present landfill methods and possible alternative disposal schemes. My findings, presented below, have been reviewed by A.C. Barlow and R.G. Kissell, Jr. (ESD Water Resources & Pollution Control) who are familiar with the Newport Plant.

SUMMARY AND CONCLUSIONS

1. Both hexavalent and trivalent chromium can be dissolved from the waste tape. Laboratory tests indicate that only a small percentage of the total tape chromium content will dissolve. The effect of long-term burial in a landfill site is unknown. Even if small amounts of chromium dissolve from the tape, the potential for ground water or stream pollution increases as larger quantities of tape are buried.
2. Carefully supervised and controlled burial of waste "Crolyn" tape at the present Newport Plant landfill site is the best alternative for the short term. Other disposal methods, preferably excluding landfill, must be developed if continued large scale production of waste tape is anticipated.
3. Borings have been made in the vicinity of the "Crolyn" tape disposal area and well points have been placed in some of these bore holes. The test borings indicate that the soil in the vicinity of the disposal area will retard migration of contaminants. However, care must be taken that tape is not buried in sandy water bearing lower layers. Water samples from the test wells show very low chromium levels which may be background concentrations.

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ORIGINAL
(Red)

2
February 10, 1970
W. B. Jennings

4. Drums containing CrO_2 or CrO_2 dispersion mixtures should not be buried at the Newport landfill site. Alternate disposal methods, such as contract disposal, or CrO_2 dust calcination for return to chromate producers should be considered.

RECOMMENDATIONS

1. Burial of the waste tape should be confined to the upper 6 or 7 feet of the landfill area. If a sandy water bearing strata is exposed in digging a new pit, this pit must not be used and should be refilled. Tape should be collected in and buried in large polyethylene bags as is presently done. Compaction of the tape to the greatest degree possible is recommended.
2. Water from the sample wells in the landfill area should be sampled quarterly and analyzed for chromate and total chromium content. Since original well samples were not large enough to get an accurate chromium analysis, new samples should be taken. These should be compared with samples from the Christina River and plant wells removed from the landfill site (this is presently underway).
3. If it is anticipated that large volumes of scrap tape will continue to be produced, alternate disposal schemes will have to be developed. I recommend that incineration and return of Cr_2O_3 ash to chromic acid manufacturers or other reuse schemes be considered rather than landfilling of large quantities of this material.

My detailed discussion is presented on the following pages.

If you have any questions or wish further assistance, please feel free to contact me.

ENGINEERING SERVICE DIVISION


P. A. Palmer

PAP:lp
Attach.

AR200041

DISCUSSION

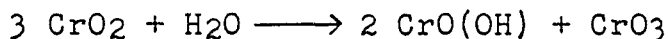
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Burial of waste "Crolyn" magnetic tape at the Newport Plant landfill site was begun in 1967. To date, the equivalent of 41,000 pounds of CrO_2 is buried at the Newport site and an additional 10,000 pounds is anticipated for 1970. These large quantities of landfilled chromium posed questions of the ultimate fate of the tape, the desirability of this method of disposal with respect to potential ground water and stream pollution, and the alternate disposal methods available. Each one of these aspects is reviewed below.

Tape Decomposition

Present knowledge of the decomposition of both CrO_2 and CrO_3 bearing tape are largely the result of Dr. W.D. Bottjer's work which I will summarize below.

Chromium dioxide (CrO_2) will slowly dissolve in water through the following disproportionation reaction:



One-third of the chromium then would appear in the hexavalent form and the other two-thirds appear in the trivalent form. Early work on distilled water extraction of chromium dioxide powder showed a rate of decomposition of CrO_2 of approximately 0.03 wt %/hr at 25°C.(1) Other work indicated that a steady state concentration of about 100 mg/l CrO_4 might be attained at room temperature.(2) Elevated temperatures (65°C) might increase this to about 500 mg/l.

Early tests indicated that the tape would generate chromium approximately 100 times less rapidly than the oxide and that as little as 0.1% of the total chromium would be available for extraction. More recent studies of the buildup of CrO_3 in distilled water in contact with various tape formulations have given CrO_3 concentrations of 0.5 - 1.3 ppm for 72 hours' exposure, 0.5 - 2.7 ppm for 168 hours' exposure, and about 3 ppm for one month's exposure.(3) These laboratory tests also indicated that 3 ppm CrO_3 might be a maximum and that some of the CrO_2 and perhaps some of the CrO_3 formed would be reduced by the binder in the tape. Although it appears that a significant amount of the CrO_2 or CrO_3 breakdown products may be permanently trapped in the tape, the fact remains that some soluble chromium, particularly the chromate ion, will be released. The effect of long-term burial on tape decomposition and production of soluble chromium compounds is unknown and probably cannot be adequately simulated in the laboratory.

We must conclude that soluble chromium compounds will be produced by landfilling.

Water Pollution Potential

The concern in burying the waste tape at the Newport landfill is that soluble chromium compounds from the landfill site will either contaminate the area ground water supply or pollute the

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Christina River. The United States Public Health Service has a mandatory limit of 0.05 mg/l (0.05 ppm) for hexavalent chromium in domestic water supplies. No limit has been set for trivalent chromium. Concentration of hexavalent chromium in ground water must then be limited to this amount if it is used as a source of drinking water. This limit is twenty times lower than the concentration of hexavalent chromium measured after prolonged exposure of tape to distilled water in the laboratory. In addition, the potential for pollution grows as the quantity of buried tape is increased. An estimated 51,000 pounds of CrO₂ in waste tape will be buried in the dump by the end of 1970. If we assume that one-third of the chromium from this tape becomes dissolved as hexavalent chromium, it would take 26 billion gallons of ground water to dilute it to the USPHS standard. Even if only .1% of this is taken as the probable amount of hexavalent chromium which will be released, this number is still a considerable 26 MM gallons. At present, there are no known wells for drinking water within the area of the plant; however, this may not always be the case.

Delaware water quality criteria presently only define in general the tolerable level of potentially toxic pollutants. The quality limits for toxic substances pertaining to the reach of the Christina River in the Newport vicinity is stated as "none in concentration harmful (synergistically or otherwise) to humans, fish, wildlife and aquatic life". The future intended uses of the river include recreation and the maintenance and propagation of fish, aquatic life and wildlife. Delaware, as well as most other states have applied the suggested levels of pollutants presented by McKee and Wolf. (4) This reference suggests maximum levels of 0.05 mg/l Cr for both trivalent or hexavalent forms.

The amount of chromium reaching the Christina River would be a function of the drainage from the dump. The net average flow of the Christina is about 200 cu ft/sec, and the minimum about 10% of that. This means that to restrict the concentration of chromium to 0.05 ppm, chromium leakage to the river would have to be restricted to about 0.22 lb/hr during minimum flow and about ten times that during average flow. As a method of comparison, if we assume that the leakage to the Christina equals the maximum 0.03 wt % decomposition of the anticipated 51,000 pounds of chromium dioxide in the dump, 3.3 lb/hr of hexavalent chromium would be released. Considering the small amount of chromium needed to exceed the criteria we must accept that river contamination through landfill drainage is possible.

Adequacy of Present Landfill

The amount of ground water or stream pollution which may result from burying the tape at Newport is a function of many variables. The effect of burial on tape decomposition is unknown. Acid soil (most probable) would speed the reduction of hexavalent chromium to the trivalent form, but also increase the solubility of the trivalent form.

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of the hexavalent form. Organic materials in the soil would also tend to reduce the hexavalent chromium. Alkaline soil would tend to reduce the solubility and mobility of trivalent chromium compounds, but would tend to stabilize the hexavalent chromates. In the final analysis, we cannot count on soil chemistry to prevent the slow solution and migration of chromium compounds from the landfill area. The major factor in preventing ground water or stream pollution is the ability of the soil to retard the migration of the soluble compounds which may be produced.

Because large quantities of chromium tape have already been landfilled at the Newport site, and alternate disposal schemes are not immediately available, I have had tests made at the disposal site to determine the adequacy of the present method. Test borings were made to measure the permeability of the soil and piezometers were placed in a number of these borings so that ground water in the area of the disposal site could be monitored for chromium content. Exhibit I attached shows the exact location and elevation of each of the borings. Exhibit II is a graphical representation of the boring logs with the locations of the well points shown. The original boring logs are also attached. These logs show that the upper 2-13 feet is composed of silty clay with layers of sand beneath. This sand is wet and determines the water table level throughout the dump. Except for the extreme northeast corner of the landfill area, the upper silty clay layer is at least 8 feet deep. Shelby tube borings were taken throughout this layer for undisturbed samples to be used in permeability tests.

Exhibit III shows the results of permeability tests, sieve analyses, liquid limit and plasticity index tests. The measured permeabilities of less than 10^{-6} cm/sec indicate that this layer is practically impervious. Liquid limit, plasticity index and sieve analyses results confirm that the upper layer is a clayey silt which would be expected to have low permeabilities. Although the permeability of the upper layer of the dump is not as low as that of pure clay, its ability to retard migration of compounds dissolved from the tape should be good.

Five well points were placed in the bore holes toward which the drainage from the landfilled area would be expected to flow. Analyses of water samples from these wells gave no indication of chromium from Well BD-4 and less than 0.1 ppm total chromium for the remaining wells, BD-5 through BD-8. These results are at the limit of analytical ability to detect chromium and are not highly accurate because small water samples were supplied for analysis. The results are encouraging in that high concentrations of chromium were not detected. However, they are in the range of USPHS limits for chromium previously mentioned. The small amount of chromium detected may actually be the background amount in the waters of this area rather than an indication of small amounts of chromium being released from the landfill area.

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Further analyses using larger samples of water are necessary. I have recommended that each of these wells be resampled and, in order to determine background quantities of chromium, that a distant well in the plant be sampled and samples be taken of the Christina River upstream and downstream of the landfill area. This program is underway.

In light of the evidence to date, I believe that closely supervised burial of waste "Crolyn" tape may be continued at this site. I recommend that burial trenches be no deeper than 6-7 feet. If a wet sandy layer is uncovered while digging, the pit must be refilled and not used for burial. The waste tape should be buried in large polyethylene bags as is presently done. Tape should be compacted as much as possible to conserve space. Borings were only taken in the immediate area of the present burial site and it is not known how extensive the area containing adequate depths of clayey soil is. In addition to close supervision of burial, I recommend that the wells be sampled quarterly to determine if there is any increase in chromium. This must be a continuing program since migration of contaminants from landfill is normally a very slow process and contamination might not be detected for years.

I believe that landfill of waste tape at the Newport dumps is an adequate solution for the short run. However, if large quantities of waste tape are to be produced year after year, a more suitable disposal method will have to be found. Very large accumulations of waste tape buried at this location, regardless of the precautions taken, constitute a very real environmental threat. In addition, the ultimate cost of landfill will be high as land which might be used for productive purposes is used up.

Alternative Disposal Methods

I have surveyed other methods which might be used for disposal of the tape if continued high scrap production is anticipated. My appraisal of various alternate methods is outlined in the following few paragraphs.

Burial at other sites in Newport is possible. Even though County landfill areas are thoroughly surveyed for potential of ground water pollution, I have ruled these out for "Crolyn" tape because of security reasons. Suitable areas for landfill are at a premium in Delaware and the situation is extremely critical in New Castle County. Contributions of large quantities of tape to these landfill areas would not be welcomed. If we were to dispose of continued large quantities of this tape by burial on Du Pont sites, I would recommend that it be done in specifically excavated pits for this purpose which would be clay-lined. This would be expensive, particularly if purchase of new land would be required.

AR200045

I have checked the possible use of land at Cherry Island presently belonging to the Pigments Department. This land is urgently needed by Pigments Department for both chemical dumping and solid waste dumping. In addition, they are already experiencing pollution problems from the waste disposal at the plant, and this area can be ruled out for disposal of "Crolyn". Generally, burial of "Crolyn" tape is unattractive from both the pollution and economic standpoint in the long run.

Incineration is attractive in that the residue ash would consist primarily of Cr_2O_3 . This material might then be returned to chromic acid manufacturers for reprocessing. Incineration of Mylar® base tapes is not easy and considerable developmental effort would be required to design a workable system. Attention will have to be given to the control of emissions of chromium containing compounds to the atmosphere. Although neither incineration nor other methods of recovery of chromium from the tape will be inexpensive, such methods for the disposal of tape must be developed if long-term production of waste tape is anticipated. Consequently, I recommend that a Photo Products Department program be instituted to find an acceptable alternate means of disposal, stressing recovery or recycle of the chromium.

Contract disposal of the tape has not looked attractive, primarily because the disposal services would haul the wastes away only to be landfilled at another location. We would have little or no control of the way the landfill is carried out and we would have problems with security in being sure that the waste was disposed of. There is a new commercial waste disposal company, Rollins-Purle, Incorporated, (5) which is constructing waste dispersal facilities in Bridgeport, New Jersey. I believe they will run a reliable, competent operation and that we should have no security problems in dealing with these people.

James McLaughlin of Rollins-Purle said that they may be able to dispose of the tape, and will submit a proposal to me for the cost of haulage and disposal.

Drums of Settled CrO_2 Dispersion

In 1968, I recommended that Atlantic Industrial Tank Maintenance, Inc. be contracted to dispose of 60 water-filled drums containing settled CrO_2 dispersion. (6) Since that time, an additional 220 drums have been disposed of by this firm. About 150-200 more drums must be disposed of in 1970. Although I am somewhat concerned about Atlantic Tank Maintenance's burial procedures in light of the larger number of drums being buried than I anticipated, I believe we can continue to use their disposal service. I am engaged in a study of disposal services in this area. If I find their method of disposal is inadequate or that there are other services which will handle these wastes more adequately, I will let you know. I have contacted Rollins-Purle about these wastes and they feel that they can dispose of them. I do not have cost information at the present time, but will advise you as more information becomes available.

AR200046

CrO₂ Powder

At present, about 10,000 pounds of off-specification CrO₂ powder is being stored at Newport. Unknown quantities of this material will be produced in the future. Some consideration had been given to adding sodium sulfite to drums of CrO₂ and landfilling them at the Newport Plant dump. As W.G. Bottjer pointed out in his letter of October 28, 1969, the proposed addition of sodium sulfite would not be large enough to complete the conversion of CrO₂ to CrO(OH). (7) In any case, I strongly urge that drums of CrO₂ not be buried in the plant dump. As the drums corrode, large quantities of both sodium sulfite and CrO₂ would be made available for leaching in the ground water without benefit of the possible retarding or minimizing effect of a tape binder on leaching.

Since the CrO₂ contains chromium in concentrated form, I recommend that this material be returned to chromic acid manufacturers for recycle. If there is concern about security, I suggest that CrO₂ be recalcined to Cr₂O₃ before being sent off plant.

P. A. Palmer
2/10/70

AR200047

REFERENCES

C. 111
(Red)

- (1) (Tape Leaching) DRD-66-IV - Laboratory Support for Magnetic Tape Venture - Reactions of CrO₂ and Organic Compounds.
(William J. Bottjer)
- (2) Memorandum - September 28, 1967 - R.W. Hendricks to D.M. Hiller
Chromate Generation from Scrap Tape.
- (3) Memorandum - October 7, 1969 - W.G. Bottjer to B. Meerkamper
Examination of Possible Contamination of Water Supplies by
Burial of "Crolyn" Tape.
- (4) "Water Quality Criteria, 2nd Edition", McKee and Wolf - The
Resources Agency of California Publication No. 3-A (1963).
- (5) Rollins-Purle, Inc.
10 W. Baltimore Avenue
Lansdowne, Pennsylvania 19050
Phone - 215-622-5005
Mr. James McLaughlin,
Director of Marketing
- (6) Letter - September 12, 1968 - P.A. Palmer to W.B. Jennings -
Photo Products Department - Newport R&D - Waste Disposal
Magnetic Tape Venture - Waste Tape Disposal.
- (7) Memorandum - October 28, 1969 - W.G. Bottjer to R.F. Bree -
CrO₂ Waste Disposal.

AR200048

CHAIN LINK FENCE

SC636 0.0 - W1622.0
SCALED FROM DRAWING 3671 SHEET #1

TB#7 5848.0 - W1652.0
GROUND ELEV. = 13.975'

TB#1 5997.5 - W1637.0
GROUND ELEV. = 10.075'

AR200049

W1600

BD-2

TB#4 5352.5 - W1671.5
GROUND ELEV. = 11.275'

TB#5 5960.0 - W1701
GROUND ELEV. = 10.275'

BD-4

TB#2 51009.5 - W1681.0
GROUND ELEV. = 8.775'

BD-7

TB#3 51017.0 - W1720.0
GROUND ELEV. = 7.475'

TB#6 5973.5 - W1738.0
GROUND ELEV. = 10.275'

BD-3

BD-1
(TO BE SURVEYED)

ELEVATIONS BASED ON USCIG DATUM
BENCH MARK ESTABLISHED ON TOP
OF PIPE SET IN CONCRETE OF TEST
BORING #4 ELEVATION = 13.255'

DATE	DESCRIPTION	APPROVED
12-50'	LOCATION OF TEST BORINGS DUPOIT PLANT LOCATED IN NEWPORT CRISTINA HUNTERD - NEW CASTLE COUNTY, DELAWARE	
SCALE	DATE	
1"=50'	DEC. 5, 1967	
DESIGNED	DRAWN	APPROVED
WAL	WAL	
WILMINGTON	VAN DENMARK & LYNCH, INC.	DELAWARE
REFERENCE	CIVIL ENGINEER & SURVEYOR	
	DRAWING NUMBER	REV'S
	11857-12469 -5	



TITLE OF PROJ. OR STUDY _____

PROJ. OR STUDY No. _____

SUBJECT Newport Waste Tape Disposal Area

WORKS _____

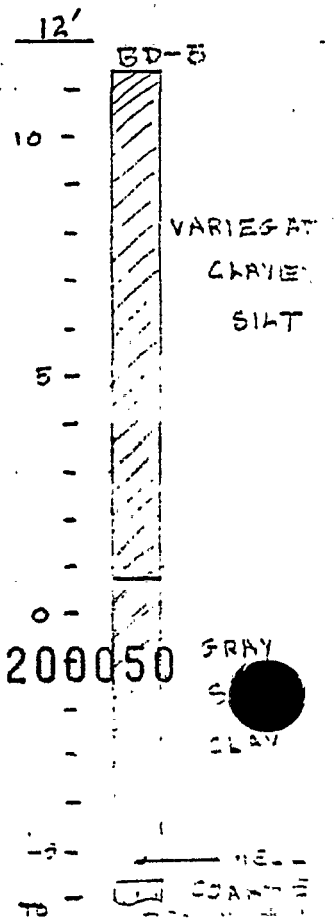
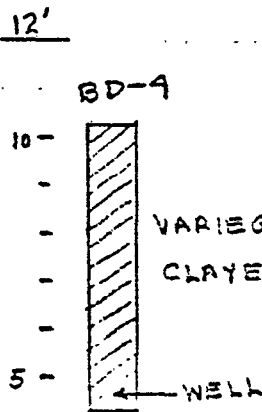
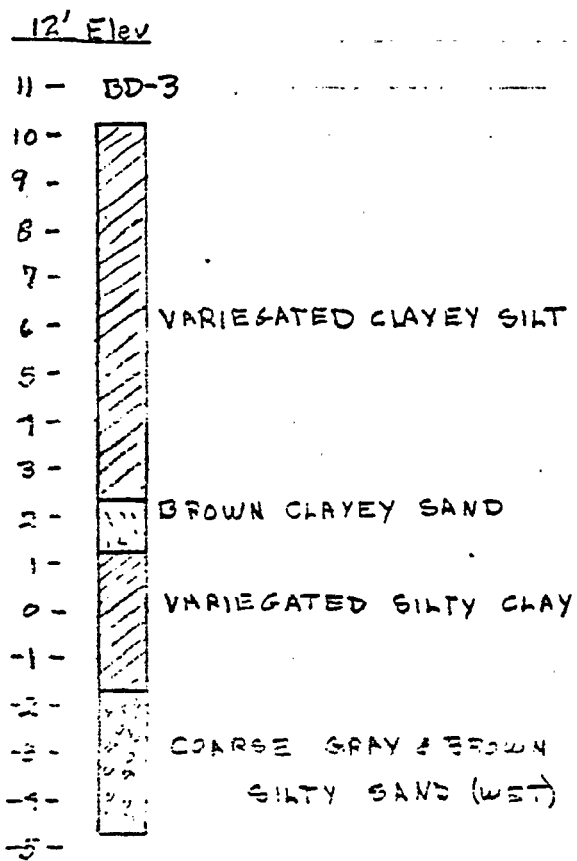
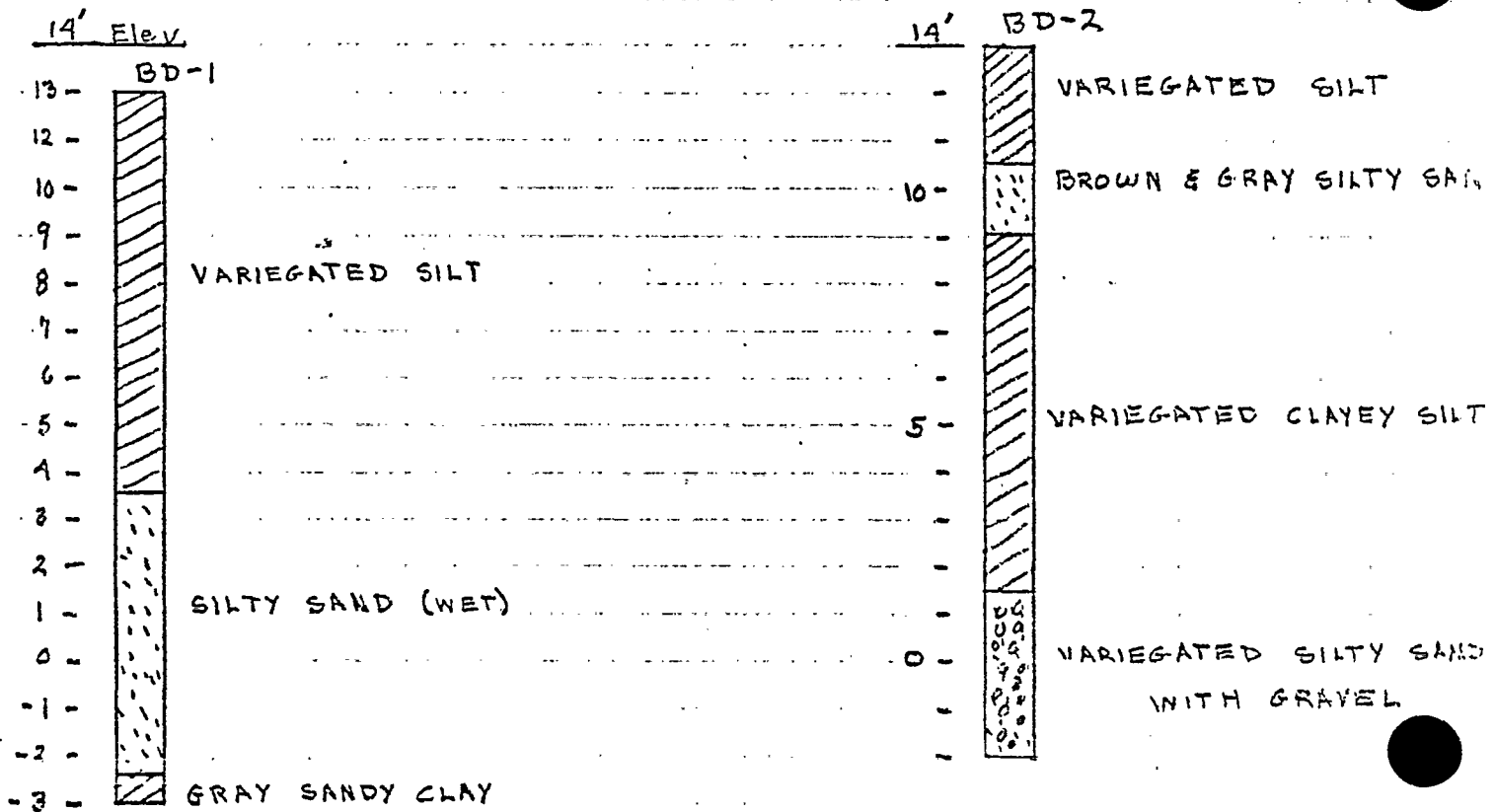
Boring logs

COMPUTER _____

P.A. Palmer

DATE _____

ORIGINAL
(2)





TITLE OF PROJ. OR STUDY _____

PROJ. OR STUDY No. _____

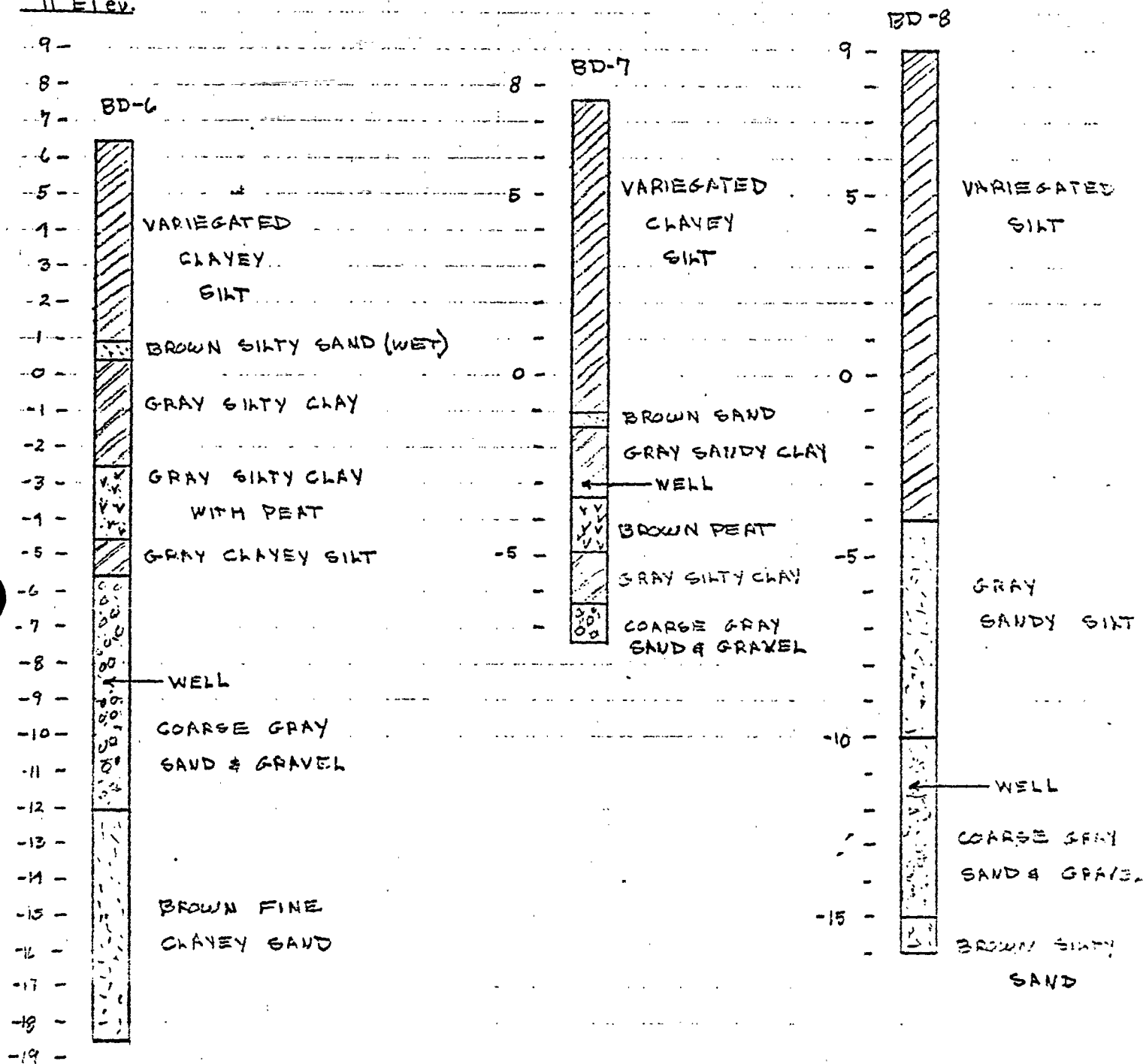
SUBJECT Waste Site Disposal Area

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Boring logsCOMPUTER P.A. Palmer

DATE _____

19/

11' Elev.

AR200051

NEWARK, DELAWARE 19711

BORING LOG

EXHIBIT II CONT

NAME DuPont Company.....PROJECT NO.
Land Fill, Newport, Delaware.....SUPERVISOR R. Howard

DRILL NO.	DRILLER	DATE
BD-1	J. Holtzman	6-13-69
ATHER	SURFACE ELEVATION	DATUM
Fair		

[illegible]

*A Number of blows of 140 lb. hammer dropped 30 in. required to drive 2 in. split-spoon sampler for each of three 6 in. increments.

* B Number of blows of 300 lb. hammer dropped 18 in. required to drive in. casing 12 inches.

REMARKS:AR200052.....

GROUND WATER

6-13-69	1:10	7.0

* BLOWNS ON CASING BAL	
0- 1	
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BORING LOG

EXHIBIT II CONT.

NAME DuPont Company.....PROJECT NO.

Inad Fill, Newport, Delaware.....SUPERVISOR R. Howard.....

BORING NO.	DRILLER	DATE
BD-6	J. Holtzman	6-12-69
WEATHER	SURFACE ELEVATION	DATUM
Fair		

[illegible]

*A Number of blows of 140 lb. hammer dropped 30 in. required to drive 2 in. split-spoon sampler for each of three 6 in. increments.

* B Number of blows of 300 lb. hammer dropped 18 in. required to drive in. casing 12 inches.

MARKS: 21.0' Gal. Pipe 2" AR200057
1 C/M Well Point (Button Type) 2"x48"

Top of Screen 19.0.....

GROUND WATER

6-13-69	1 P.M.	1.2

* BLOWS ON CASING B
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ORIGINAL
(Red)GEOTEC ASSOCIATES302 Beverly Road
Newark, DelawareNEWPORT, DELAWARE

SAMPLE	PERMEABILITY cm/sec x 10 ⁻⁷	NATURAL MOISTURE	LL	PL	PI	%Passing #20
BD 2, 4-6'	2.9	22.5	48.4	28.7	20	
3, 7-9'	72.1	15.6	NP	NP		18
5, 4-6'	16.4	24.5	28.1	23.7	4	
6, 4-5.5'	5.5	18.7	33.4	20.4	13	
7, 2-4	13.5	21.5	29.5	23.0	6	88
7, 7-8.5	6.1	21.9	27.8	23.0	5	82
8, 4-6	19.2	22.0	27.9	23.6	4	89

R. Nicholls

6-18-69

AR200060

UNIFIED SOIL CLASSIFICATION INCLUDING IDENTIFICATION AND DESCRIPTION

FIELD IDENTIFICATION PROCEDURES (Excluding particles larger than 3 inches and testing fractions on estimated weights)			GROUP SYMBOLS	TYPICAL NAMES	INFORMATION REQUIRED FOR DESCRIBING SOILS	LABORATORY CLASSIFICATION CRITERIA	
FIELD IDENTIFICATION PROCEDURES (Excluding particles larger than 3 inches and testing fractions on estimated weights)						LABORATORY CLASSIFICATION CRITERIA	
GRAVELS (More than half of coarse fraction is larger than No. 200 sieve size)	CLEAN GRAVELS (Little or no fines)	GRAVELS WITH FINES (Amount of fines)	GW	Well graded gravels, gravel-sand mixtures, little or no fines.	One typical name, indicate approximate percentages of sand and gravel, maximum size, angularity, surface condition, and hardness of the coarse grains, local or geologic name and other pertinent descriptive information, and symbol in parentheses.	$C_u = \frac{D_{60}}{D_{10}}$ Greater than 4 $C_c = \frac{(D_{30})^2}{D_{10} D_{60}}$ Between one and 3	
			GP	Poorly graded gravels, gravel-sand mixtures, little or no fines.	Not meeting all gradation requirements for GW		
SANDS (More than half of coarse fraction is smaller than No. 4 sieve size)	CLEAN SANDS (Little or no fines)	SANDS WITH FINES (Amount of fines)	GM	Silty gravels, poorly graded gravel-sand mixtures.	For undisturbed soils add information on stratification, degree of compaction, cementation, moisture conditions and drainage characteristics.	Afterberg limits below X' line, or PI less than 4	
			GC	Clayey gravels, poorly graded gravel-sand mixtures.	Afterberg limits above X' line with PI greater than 7		
FINE GRAINED SOILS (More than half of material is larger than No. 200 sieve size)	SILTS AND CLAYS (Liquid limit less than 50)	SILTS AND CLAYS (Liquid limit greater than 50)	SW	Well graded sands, gravelly sands, little or no fines.	EXAMPLE: Silty sand, gravelly, about 20% hard, angular gravel particles, 1/4 in. maximum size, rounded and subangular sand grain, coarse to fine, about 15% non-plastic fines with low dry strength, well compacted and moist in place; alluvial sand, (SM)	$C_u = \frac{D_{60}}{D_{10}}$ Greater than 6 $C_c = \frac{(D_{30})^2}{D_{10} D_{60}}$ Between one and 3	
			SP	Poorly graded sands, gravelly sands, little or no fines.	Not meeting all gradation requirements for SW		
FINE GRAINED SOILS (More than half of material is larger than No. 200 sieve size)	SILTS AND CLAYS (Liquid limit less than 50)	SILTS AND CLAYS (Liquid limit greater than 50)	SM	Silty sands, poorly graded sand-silt mixtures.	EXAMPLE: Clayey silt, brown, slightly plastic; small percentage of fine sand; numerous vertical root holes; firm and dry in place, (ML)	Afterberg limits below X' line or PI less than 4	
			SC	Clayey sands, poorly graded sand-clay mixtures.	Afterberg limits above X' line with PI greater than 7		
FINE GRAINED SOILS (More than half of material is larger than No. 200 sieve size)	SILTS AND CLAYS (Liquid limit less than 50)	SILTS AND CLAYS (Liquid limit greater than 50)	ML	Inorganic silts and very fine sand, rock flour, silty or clayey fine sand with slight plasticity.	One typical name, indicate degree and character of plasticity, amount and maximum size of coarse grains, color in wet condition, odor if any, local or geologic name, and other pertinent descriptive information, and symbol in parentheses.	$C_u = \frac{D_{60}}{D_{10}}$ Greater than 4 $C_c = \frac{(D_{30})^2}{D_{10} D_{60}}$ Between one and 3	
			CL	Organic silts and organic silt-clays of low plasticity.	For undisturbed soils add information on structure, stratification, consistency in undisturbed and remolded states, moisture and drainage conditions.	Afterberg limits below X' line or PI less than 4	
FINE GRAINED SOILS (More than half of material is larger than No. 200 sieve size)	SILTS AND CLAYS (Liquid limit less than 50)	SILTS AND CLAYS (Liquid limit greater than 50)	GM	Inorganic silts, mucous or detritaceous fine sandy or silty soils, elastic silts.	EXAMPLE: Clays silty, brown, slightly plastic; small percentage of fine sand; numerous vertical root holes; firm and dry in place, (ML)	$C_u = \frac{D_{60}}{D_{10}}$ Greater than 4 $C_c = \frac{(D_{30})^2}{D_{10} D_{60}}$ Between one and 3	
			CH	Inorganic clays of high plasticity, fat clays.	Not meeting all gradation requirements for GM		
FINE GRAINED SOILS (More than half of material is larger than No. 200 sieve size)	SILTS AND CLAYS (Liquid limit less than 50)	SILTS AND CLAYS (Liquid limit greater than 50)	OH	Organic clays of medium to high plasticity.	EXAMPLE: Clays silty, brown, slightly plastic; small percentage of fine sand; numerous vertical root holes; firm and dry in place, (ML)	Afterberg limits below X' line or PI less than 4	
			PT	Peat and other highly organic soils.	Not meeting all gradation requirements for OH		

Use grain size curve in identifying the fractions as given under field identification

Depending on percentage of gravel and sand from grain size curve, sieve size) coarse grained soils are classified as follows:

Less than 5% Gravel, 5% to 12% Gravel, More than 12% Gravel

Not meeting all gradation requirements for SW

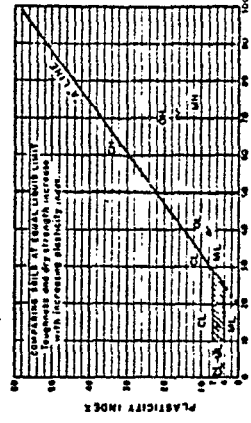
Afterberg limits below X' line or PI less than 4

Afterberg limits above X' line with PI greater than 7

Afterberg limits above X' line with PI greater than 7

PLASTICITY CHART

FOR LABORATORY CLASSIFICATION OF FINE GRAINED SOILS



PLASTICITY CHART FOR LABORATORY CLASSIFICATION OF FINE GRAINED SOILS

FIELD IDENTIFICATION PROCEDURES FOR FINE GRAINED SOILS ON FRACTIONS

These procedures are to be performed on the minus No. 40 sieve size particles, approximately 0.075 mm. For field classification purposes, screening is not intended, simply remove by hand the coarse particles that interfere with the tests.

DRY STRENGTH (Crushing characteristic)

After removing particles larger than No. 40 sieve size, mold a ball of soil in the palm of the hand. If the soil is moist, dry it in the sun or in a dry place. Allow the soil to dry completely, then break it apart with the fingers. If the soil is dry, it will break apart easily. If the soil is moist, it will break apart with difficulty. If the soil is very moist, it will not break apart.

LIQUIDITY (Reaction to shaking)

After removing particles larger than No. 40 sieve size, place a pot of moist soil in a shallow dish. Add enough water to make the soil very moist. Shake the soil with the fingers. If the soil is very moist, it will become liquid. If the soil is moist, it will become plastic. If the soil is dry, it will become solid.

TOUGHNESS (Consistency near plastic limit)

After removing particles larger than No. 40 sieve size, a specimen of soil about one-half inch cube in size is molded to the consistency of putty. If too dry, water must be added and if sticky, the specimen should be spread in a thin layer and allowed to lose some moisture by evaporation. Then the specimen is rolled out by hand on a smooth surface to a thickness of about 1/8 inch. During the rolling, the specimen should be rolled in one direction only. The specimen is then divided into three equal parts. The first part is rolled into a thread about 1/16 inch in diameter. The second part is rolled into a thread about 1/32 inch in diameter. The third part is rolled into a thread about 1/64 inch in diameter. The threads are then compared for toughness. The thread that is the stiffest, least plastic, is the toughest. The thread that is the most plastic, is the least tough.

COEFFICIENT OF PERMEABILITY CHART BY A. CASAGRANDE

"k" in cm. per sec. $\times 1.97 =$ ft. per min. or $\times 14.75 =$ gal. per sq. ft. per min. (log scale)

10⁻² 10⁻¹ 1.0 10⁻¹ 10⁻² 10⁻³ 10⁻⁴ 10⁻⁵ 10⁻⁶ 10⁻⁷ 10⁻⁸ 10⁻⁹

RAINAGE PROPERTIES	Good Drainage	Poor Drainage	Practically Impervious
LOCATION WITH DAMS AND DIKES	Pervious Sections of Earth Dams and Dikes	Impervious Sections of Earth Dams and Dikes	
ES OF SOIL	Clean Gravel Clean Sand Clean Sand and Gravel Mixtures	Very Fine Sands; Organic and Inorganic Silts; Mixtures of Sand, Silt, and Clay; Glacial Till; Stratified Clay Deposits; etc.	"Impervious" Soils e.g. Homogeneous Clays Below Zone of Weathering
DIRECT TERMINATION COEFFICIENT PERMEABILITY	DIRECT TESTING OF SOIL IN ITS ORIGINAL POSITION (e.g. Well Points) Reliable if properly conducted. Considerable Experience Required.	"Impervious Soils" which are modified by the effects of vegetation and weathering	
	CONSTANT HEAD PERMEAMETER Little Experience Required		
	TURBULENT FLOW FOR HYDRAULIC GRADIENTS LARGER THAN TEN.	FALLING HEAD PERMEAMETER Unreliable Much Experience Necessary for Correct Interpretation	Fairly Reliable Considerable Experience Necessary
INDIRECT TERMINATION COEFFICIENT PERMEABILITY	COMPUTATIONS from Grain Size Distribution. (e.g. Hazen's Formula). Applicable Only to Clean Cohesionless Sands and Gravels		
		HORIZONTAL CAPILLARITY TEST Very Little Experience Necessary. Especially Useful for Rapid Testing of a Large Number of Samples in the Field without Laboratory Facilities.	COMPUTATIONS From Consolidation Tests. Expensive Laboratory Equipment and Considerable Experience Required.

AR200062

ORIGINAL

GEOTEC ASSOCIATES302 Beverly Road
Newark, DelawareORIGINAL
(Recd)NEWPORT

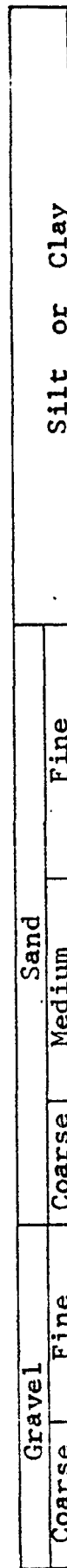
Boring	Sample	Natural Moisture	LL	PL	PI	% Passing #200
BD 5	1	38.1	29.7	25.3	4	79
	3	23.6	28.1	21.5	7	89
	4	22.7	25.3	20.1	5	87
	5	31.4	29.4	23.2	6	99
	6	32.5	33.6	24.0	10	96
	7	30.8	NP	NP		see attached sieve analysis
	8	23.5	NP	NP		sieve
BD6	1	21.0	30.4	20.8	10	88
	2	14.4	28.5	19.9	9	91
	4	20.3	20.8	18.0	3	65
	5	118	55.6	43.0	13	100
	6	52.1	42.1	32.8	9	96
	7	15.5	NP	NP		sieve
	8	16.7	NP	NP		sieve
	9	13.5	NP	NP		sieve

AR200063

6-25-69

rln

302 Beverly Road
Newark, Delaware



Location	Sample	Depth	Nat W %	LL	PL	PI
Newport BD 5	7		30.8	NP	NP	
	8		23.5	NP	NP	

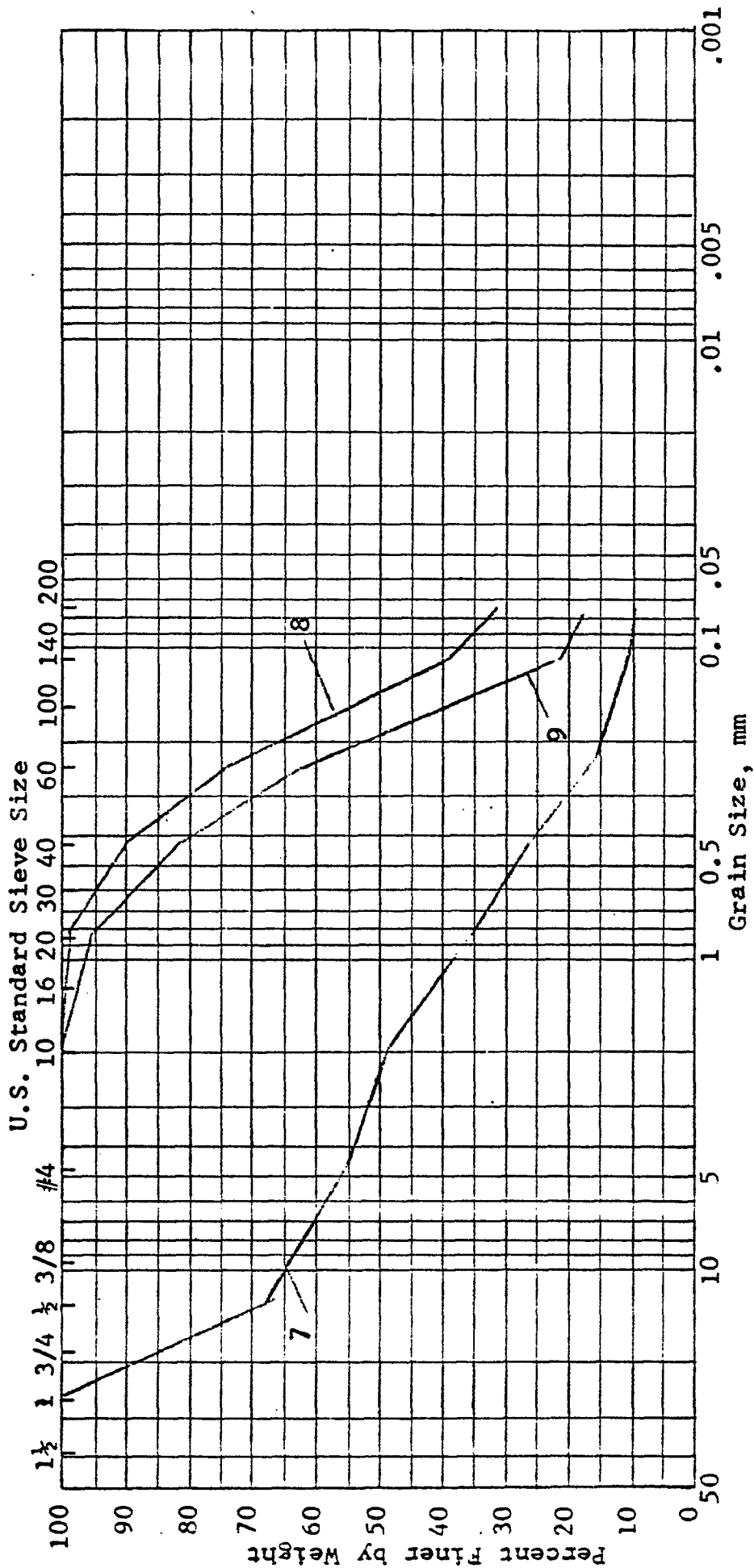
Technician
rln

Date 6--25-69

SOIL CLASSIFICATION TESTS

GEOTEC ASSOCIATES

302 Beverly Road
Newark, Delaware



Gravel		Sand		Silt or Clay	
Coarse	Fine	Coarse	Medium	Fine	

AP 2000065

Location	Sample	Depth	Nat W %	LL	PL	PI
Newport BD 6	7		15.5	NP	NP	
	8		16.7	NP	NP	
	9		13.5	NP	NP	

Technician rln

Date 6-25-69

SOIL CLASSIFICATION TESTS

Original
(Red)

Mike

Land-CO₂ Disposal
History

Limited details to date obtained from M. Stuyffer

- Some buried under #95 as extra fill (Believed) (Before 76)
- Some buried in an old Edgemoor landfill (Before 76) Believed
- 1976-1979 Collins-Bridgport N.J. Landfill
- 1979-1981 Stored on George Huber's site near Edgemoor
- Summer 81 - Some stored on George Huber's site
 - 95% sent to BFI - Landfill - Kelly, Rock-
Bottoms, Md.

Pete

AR200066



E. I. DU PONT DE NEMOURS & COMPANY
INCORPORATED
WILMINGTON, DELAWARE 19898

PHOTO PRODUCTS DEPARTMENT

ORIGINAL
(Red)

December 10, 1969

TO: P. A. PALMER, L-13 WIO

FROM: W. B. JENNINGS *WBJ*

C O N F I D E N T I A L 

SCRAP TAPE QUANTITIES

The following quantities of magnetic tape, expressed in terms of CrO_2 contained, have been buried at the Newport Plant Dump.

<u>YEAR</u>	<u>M LBS. EQUIVALENT CR O₂</u>
1967	2.5
1968	12.5
1969	26.0
TOTAL	41.0
Estimate 1970	10.0

WBJ/jhb

cc: R. M. Haire

AR200067

Photo Products Department
Experimental Station

cc: ~~R. Waller-Engg., LION~~ ^{ORIGINAL}
W. B. Jennings-Newport
B. Meerkamper-E356 (red)
J. E. Dickens-E356
File A-4.2

September 28, 1967

MEMORANDUM

TO: D. M. HILLER

FROM: R. W. HENDRICKS *RW Hendricks*

CHROMATE GENERATION FROM SCRAP TAPE

Reference: Your Speedimemo of September 14, 1967

You requested information on the probability that scrap "Crolyn"TM magnetic tape would generate chromates when weathered, e.g., on the Newport plant dump and, hence, constitute a pollution problem.

We have information on a tape sample, H-13D (an early Huntley J65 formulation tape), which has been soaked in water for two years, which indicates the pollution hazard from scrap tape is minimal, if indeed there is any hazard. We do not feel this problem need be a cause for concern.

The work which follows is described on pages 223 and 227 of my notebook 65-II. On August 30, 1965, a 25 square inch sample of tape H-13D (containing approximately 214 mg. CrO_2) was put in a 4-ounce bottle of distilled water (about 115 ml.). A second bottle was made up with 200 mg. of CrO_2 D8-157 in 4 ounces of water. After 24 hours, no color was visible in the tape bottle, while the bottle containing oxide had reached a chromate concentration of approximately 1 mg./l. (from visual comparison with standard solutions). After two weeks (September 14, 1965), there was still no visible color in the tape bottle, and a test by W. G. Bottjer for chromate was negative. (From other work, 65-II, page 245, the oxide bottle would have been at a chromate level of approximately 50-100 mg./liter at this point. The two bottles were next placed in a 65°C. oven for twenty hours, after which the water in the tape bottle appeared "very slightly yellow," and the water in the oxide bottle had increased to a color corresponding to approximately 500 mg. CrO_4 per liter.

AR200068

ORIGINAL
(Red)

(Unfortunately, the contents were not analyzed more accurately at that time.) The bottles were then left to stand in John Dickens' office until now, and both appear to be unchanged after two years. The bottle containing tape still appears very slightly colored, and by comparison with standard solutions, the water contains approximately 5 mg. CrO_4 per liter. The bottle containing oxide still has about 500 mg. per liter, again by comparison with standards.

Despite the very slight water coloration which appears when tape is exposed to water even at 65°C ., magnetic degradation is severe. Work by H. G. Ingersoll in later 1965 (my notebook 65-III, October 5, 1965) showed that a typical J365 tape lost 13% magnetically in 24 hours in water, versus 7% in air at 65°C ./50% RH. The inference is that the oxide (or the chromic acid resulting from hydrolysis) reacts almost entirely with the binder, even when tape is soaked in water. Significantly, we have never found free chromic acid in oxide which has been exposed to a reducing agent, or from degraded tape samples. (W. G. Bottjer made careful x-ray studies in late 1965. The only degradation product found was $\gamma\text{-CrOOH}$.)

From this, we conclude that tape stored outdoors, or in contact with a large excess of water at room temperature, would generate essentially no extractable chromates. Tape generates chromates at least one hundred times less rapidly than oxide at 65°C . In my experiment, all of the 5 mg. per liter apparently originated during the 65° cycle, and virtually none at room temperature in two years. (No more than about 0.1% of the available chromium.)

If this does not suffice for your purposes, I suggest that larger quantities of tape be soaked in a large excess of water at room temperature, and the chromate concentration of the water be monitored by looking for color buildup. Standard colorimetric comparison methods can be used to establish low chromate levels quite accurately. This approach would, of course, actually represent an accelerated test, since tape would actually not be constantly soaked in liquid water on the Newport dump. This might, however, suffice to establish an upper limit of extractables more accurately than my data permit.

Call me if further clarification is needed.

RWH/dlc
9/28/67

AR200069

ORIGINAL
(Red)

E. I. DU PONT DE NEMOURS & COMPANY
PIGMENTS DEPARTMENT

CC: E. L. Rodowskas
R. M. Salemi
E. H. Kolb
W. A. Roberson

Newport, Delaware
May 31, 1968

MEMORANDUM

TO: ~~W. B. JENNINGS~~ (2)
FROM: W. E. STUEFER *MD*

Tape contaminated with chromium dioxide, as you have frequently indicated, is a toxicity problem if burned. It is also a handling problem due to bulk and lightness. In accord with the need to keep separate from flammables and to minimize the burying problem, it is requested that all tape of this nature be kept separate from other area scrap and that it be bagged or secured in some other manner. It is also requested that means of further compacting it in the future be investigated.

WES/fc

AR200070

PHOTO PRODUCTS DEPARTMENT
Newport Plant

CC: R. M. Salemi
E. H. Kolb
W. B. Jennings III
M. Chiaverini
P. S. duPont IV
L. S. Smith

ORIGINAL
(Red)

May 31, 1968

TO: W. E. STUEFER

FROM: J. R. HOGAN

Dump Materials -- Forecast

Ref: Letter to EHK from JRH -- 5/14/68

The following is additional information that you requested:

A. Dempster Dumpster Loads -- Dispersion/Coater Materials

<u>Material Volumes</u>	<u>Forecast -- Date</u>		
	<u>Present</u>	<u>3Q68</u>	<u>4Q68</u>
1. Scrap Magnetic Tape, ft. ³	1650	3500	7000
2. Compar Tubing Containing Dried Dispersion, ft. ³	25	25	25
3. Glass Jars and Tubing, ft. ³	2	3	5
4. Tin Cans, ft. ³	10	15	25
5. Small Amounts of Dispersion, ft. ³	1	1	2
6. Spent Filterite and Rigimesh filters, ft. ³	9	12	30
7. Lever Paks, ft. ³	80	80	120
TOTAL VOLUME, Ft. ³	1777	3636	7207
Number of Dempster Dumpster Loads, Per Month	11	22	44

AR200071

5/31/68

PHOTO PRODUCTS DEPARTMENT
Newport Plant

ORIGINAL
(Red)

file
CC: E. L. Rodowskas
R. M. Salemi
E. H. Kolb
L. S. Smith
M. Chiaverini
Dr. Waller

June 28, 1968

TO: W. E. STUEFER

FROM: W. B. JENNINGS

DISPOSAL OF SCRAP MAGNETIC TAPE

(Ref. your letter of 5/31/68 to W. B. Jennings)
(Ref. your letter of 6/11/68 to L. S. Smith)

Your letter of 5/31/68 requests more complete segregation of chromium containing waste from the Magnetic Tape area to permit proper disposal at the plant dump. While such segregation has been our previous practice, we have re-emphasized this fact to all personnel and we will follow-up to insure compliance. Our practice is summarized in the attached letter from M. Chiaverini.

We have investigated the possible use of a baler to reduce the bulk volume of scrap tape. High spot cost is \$5-6000 without building.

Your letter of 6/11/68 requests that we consider off-plant disposal locations for chromium containing waste. We have requested the assistance of Dr. Robert Waller in searching for such locations. We will keep you informed of progress.

WBJ:evb
Attach.

AR200074

8/1/68

CC: W. B. Jennings III
L. S. Smith
ORIGINAL
(Red)

June 13, 1968

TO: ALL MAGNETIC TAPE PERSONNEL

FROM: MAURO CHIAVERINI

DISPOSAL OF WASTE MATERIALS AND HOUSEKEEPING OF OUTSIDE AREAS

To comply with Newport Plant waste disposal regulations and to maintain a high level of housekeeping standard in our "outside" areas these procedures must be followed:

1. Cigarette butt cans located in "outside" areas are to be used for disposing of cigarette butts. Papers and other waste should not be deposited in these cans.
2. Efforts are necessary in separating materials going to the Newport Plant dump from MDF. All MDF scrap materials are to be separated into three types of scrap. These are:
 - a. All scrap magnetic tape is to be put in polyethylene bags to prevent the tape from being scattered over the outside areas. These bags of tapes are to be disposed of in the Dempster Dumpster marked for "Trash Tape Only".
 - b. All items that have come in contact with CrO₂ dispersion and/or solvent will be disposed of in empty Lever Paks located in a designated roped-off area on the south side of A-212. Examples of these items are: Texwipes and Nainsook rags used for cleaning up dispersion. (These items should have been deposited in red safety trash cans prior to their being removed to the lever paks; 2. spent filter elements; and 3. small amounts of dispersion, etc.
 - c. All other items are to be disposed of in the Dempster Dumpster marked for "Trash Only". Examples of these items are:
 1. cardboard boxes and other paper product items;
 2. wood items, etc.


The Dempster Dumpsters are to be taken to the Plant dump every scheduled working day. The lever paks are to be taken to the dump, preferably after each shift, but should depend on the fullness of them.

AR200075

Parking of cars in such a manner than blocks access to removal of the Dempster Dumpster to the Plant dump must be discontinued.

PHOTO PRODUCTS DEPARTMENT
Newport Plant

CC: E. L. Rodowskas
R. M. Salemi
E. H. Kolb
L. S. Smith
M. Chiaverini
Dr. Waller



ORIGINAL
(Red)

June 28, 1968

TO: W. E. STUEFER

FROM: W. B. JENNINGS

DISPOSAL OF SCRAP MAGNETIC TAPE

(Ref. your letter of 5/31/68 to W. B. Jennings)
(Ref. your letter of 6/11/68 to L. S. Smith)

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WBJ:evb
Attach.

AR200076

8/1/68

CC: W. B. Jennings III
L. S. Smith

June 13, 1968

TO: ALL MAGNETIC TAPE PERSONNEL

FROM: MAURO CHIAVERINI

DISPOSAL OF WASTE MATERIALS AND HOUSEKEEPING OF OUTSIDE AREAS

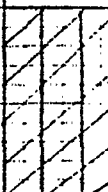

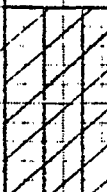
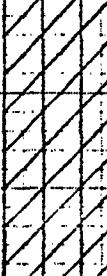
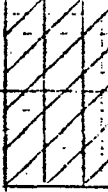
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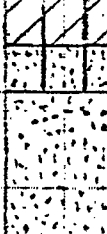
The Dempster Dumpsters are to be taken to the Plant dump every scheduled working day. The lever paks are to be taken to the dump immediately after each shift, but should depend on the fullness of them.

Parking of cars in such a manner that blocks access to removal of the Dempster Dumpster to the Plant dump must be discontinued.

AR200077

				LOG OF AUGER BORROW PIT - EXPLORATION NEWPORT PROJ. 9720N FEB. 23, 1953 H. R. SLANEY, JR. SHEET 1 OF 2		
24	1-B S 380° W 1640°					
23						
22	TOP SOIL					
21		INTERBEDDED YELLOW BROWN & GRAY CLAYEY SILT		2-B S 380° W 1820°		
20				3-B S 380° W 2030°		
19				TOP SOIL		
18		VERY HARD			TOP SOIL	
17			INTERBEDDED GRAY & BROWN SILTY CLAY WATER TABLE			
16			INTERBEDDED BROWN, YELLOW & GRAY CLAYEY SILT			
20				5-B S 580° W 1985°		
19				TOP SOIL	6-B S 1120° W 2000°	
18					TOP SOIL	
17	4-B S 580° W 2030°				REDDISH BROWN CLAYEY SILT GRADING TO GRAY	REDDISH CLAYEY SILT MICA PARTICLES
16						
15	TOP SOIL					
14		INTERBEDDED BROWN & YELLOW CLAYEY SILT VERY HARD				
13						
12						

AR200078



AR200078





ORIGINAL
(Red)

	7-B S 1360± W 1750±		8-B S 1200± W 2100±		9-B ✓ S 1080± W 1940±	
10		TOPSOIL		TOPSOIL		TOPSOIL
9		INTERBEDDED REDDISH-		REDDISH-BROWN CLAYEY SILT		INTERBEDDED REDDISH-
8		BROWN		REDDISH-BROWN CLAYEY SANDY SILT		BROWN &
7		GRAY		GRAYISH-BROWN SANDY SILT		GRAY
6		CLAYEY SILT		BROWN SAND & SMALL STONES		CLAYEY SILT
5				GRAYISH-BROWN SANDY SILT		
		WATER TABLE		BROWN SAND & SMALL STONES		
			11-B S 950± W 2100±		12-B S 750± W 2150±	
16				TOPSOIL		TOPSOIL
15	10-B S 1080± W 1940± ✓					INTERBEDDED BROWN &
14				INTERBEDDED BROWN &		GRAY
		TOPSOIL		GRAY		CLAYEY SILT
13				CLAYEY SILT		VERY HARD
12		INTERBEDDED REDDISH-				
		BROWN &				
11		GRAY				
		CLAYEY				
10		SILT				

LOG OF AUGER BORINGS
BORROW PIT EXPLORATION
NEWPORT
PRCL 7720-IV
FEB. 23, 1953
H. P. SLANEY, JR.
SHEET

AR200079